

FHSST Authors

**The Free High School Science Texts:
Textbooks for High School Students
Studying the Sciences
Physics
Grades 10 - 12**

**Version 0
November 9, 2008**

Copyright 2007 "Free High School Science Texts"

Permission **is** granted to copy, distribute and/or modify this document under the terms of the GNU Free Documentation License, Version 1.2 or any later version published by the Free Software Foundation; with no Invariant Sections, no Front-Cover Texts, and no Back-Cover Texts. A copy of the license is included in the section entitled "GNU Free Documentation License".

STOP!!!!

Did you notice the **FREEDOMS** we've granted you?

Our copyright license is **different!** It grants freedoms rather than just imposing restrictions like all those other textbooks you probably own or use.

- We know people copy textbooks illegally but we would LOVE it if you copied our's - go ahead copy to your hearts content, **legally!**
- Publishers' revenue is generated by controlling the market, we don't want any money, go ahead, distribute our books far and wide - we DARE you!
- Ever wanted to change your textbook? Of course you have! Go ahead, change ours, make your own version, get your friends together, rip it apart and put it back together the way you like it. That's what we really want!
- Copy, modify, adapt, enhance, share, critique, adore, and contextualise. Do it all, do it with your colleagues, your friends, or alone but get involved! Together we can overcome the challenges our complex and diverse country presents.
- So what is the catch? The only thing you can't do is take this book, make a few changes and then tell others that they can't do the same with your changes. It's share and share-alike and we know you'll agree that is only fair.
- These books were written by volunteers who want to help support education, who want the facts to be freely available for teachers to copy, adapt and re-use. Thousands of hours went into making them and they are a gift to everyone in the education community.

FHSST Core Team

Mark Horner ; Samuel Halliday ; Sarah Blyth ; Rory Adams ; Spencer Wheaton

FHSST Editors

Jayne Padayachee ; Joanne Boulle ; Diana Mulcahy ; Annette Nell ; René Toerien ; Donovan Whitfield

FHSST Contributors

Rory Adams ; Prashant Arora ; Richard Baxter ; Dr. Sarah Blyth ; Sebastian Bodenstein ; Graeme Broster ; Richard Case ; Brett Cocks ; Tim Crombie ; Dr. Anne Dabrowski ; Laura Daniels ; Sean Dobbs ; Fernando Durrell ; Dr. Dan Dwyer ; Frans van Eeden ; Giovanni Franzoni ; Ingrid von Glehn ; Tamara von Glehn ; Lindsay Glesener ; Dr. Vanessa Godfrey ; Dr. Johan Gonzalez ; Hemant Gopal ; Umeshree Govender ; Heather Gray ; Lynn Greeff ; Dr. Tom Gutierrez ; Brooke Haag ; Kate Hadley ; Dr. Sam Halliday ; Asheena Hanuman ; Neil Hart ; Nicholas Hatcher ; Dr. Mark Horner ; Robert Hovden ; Mfandaidza Hove ; Jennifer Hsieh ; Clare Johnson ; Luke Jordan ; Tana Joseph ; Dr. Jennifer Klay ; Lara Kruger ; Sihle Kubheka ; Andrew Kubik ; Dr. Marco van Leeuwen ; Dr. Anton Machacek ; Dr. Komal Maheshwari ; Kosma von Maltitz ; Nicole Masureik ; John Mathew ; JoEllen McBride ; Nikolai Meures ; Riana Meyer ; Jenny Miller ; Abdul Mirza ; Asogan Moodaly ; Jothi Moodley ; Nolene Naidu ; Tyrone Negus ; Thomas O'Donnell ; Dr. Markus Oldenburg ; Dr. Jaynie Padayachee ; Nicolette Pekeur ; Sirika Pillay ; Jacques Plaut ; Andrea Prinsloo ; Joseph Raimondo ; Sanya Rajani ; Prof. Sergey Rakityansky ; Alastair Ramlakan ; Razvan Remsing ; Max Richter ; Sean Riddle ; Evan Robinson ; Dr. Andrew Rose ; Bianca Ruddy ; Katie Russell ; Duncan Scott ; Helen Seals ; Ian Sherratt ; Roger Sieloff ; Bradley Smith ; Greg Solomon ; Mike Stringer ; Shen Tian ; Robert Torregrosa ; Jimmy Tseng ; Helen Waugh ; Dr. Dawn Webber ; Michelle Wen ; Dr. Alexander Wetzler ; Dr. Spencer Wheaton ; Vivian White ; Dr. Gerald Wigger ; Harry Wiggins ; Wendy Williams ; Julie Wilson ; Andrew Wood ; Emma Wormauld ; Sahal Yacoob ; Jean Youssef

Contributors and editors have made a sincere effort to produce an accurate and useful resource. Should you have suggestions, find mistakes or be prepared to donate material for inclusion, please don't hesitate to contact us. We intend to work with all who are willing to help make this a continuously evolving resource!

Contents

I Introduction	1
1 What is Physics?	3
II Grade 10 - Physics	5
2 Units	9
2.1 Introduction	9
2.2 Unit Systems	9
2.2.1 SI Units	9
2.2.2 The Other Systems of Units	10
2.3 Writing Units as Words or Symbols	10
2.4 Combinations of SI Base Units	12
2.5 Rounding, Scientific Notation and Significant Figures	12
2.5.1 Rounding Off	12
2.5.2 Error Margins	13
2.5.3 Scientific Notation	13
2.5.4 Significant Figures	15
2.6 Prefixes of Base Units	15
2.7 The Importance of Units	17
2.8 How to Change Units	17
2.8.1 Two other useful conversions	19
2.9 A sanity test	19
2.10 Summary	19
2.11 End of Chapter Exercises	21
3 Motion in One Dimension - Grade 10	23
3.1 Introduction	23
3.2 Reference Point, Frame of Reference and Position	23
3.2.1 Frames of Reference	23
3.2.2 Position	25
3.3 Displacement and Distance	28
3.3.1 Interpreting Direction	29
3.3.2 Differences between Distance and Displacement	29
3.4 Speed, Average Velocity and Instantaneous Velocity	31

3.4.1 Differences between Speed and Velocity	35
3.5 Acceleration	38
3.6 Description of Motion	39
3.6.1 Stationary Object	40
3.6.2 Motion at Constant Velocity	41
3.6.3 Motion at Constant Acceleration	46
3.7 Summary of Graphs	48
3.8 Worked Examples	49
3.9 Equations of Motion	54
3.9.1 Finding the Equations of Motion	54
3.10 Applications in the Real-World	59
3.11 Summary	61
3.12 End of Chapter Exercises: Motion in One Dimension	62
4 Gravity and Mechanical Energy - Grade 10	67
4.1 Weight	67
4.1.1 Differences between Mass and Weight	68
4.2 Acceleration due to Gravity	69
4.2.1 Gravitational Fields	69
4.2.2 Free fall	69
4.3 Potential Energy	73
4.4 Kinetic Energy	75
4.4.1 Checking units	77
4.5 Mechanical Energy	78
4.5.1 Conservation of Mechanical Energy	78
4.5.2 Using the Law of Conservation of Energy	79
4.6 Energy graphs	82
4.7 Summary	83
4.8 End of Chapter Exercises: Gravity and Mechanical Energy	84
5 Transverse Pulses - Grade 10	87
5.1 Introduction	87
5.2 What is a <i>medium</i> ?	87
5.3 What is a <i>pulse</i> ?	87
5.3.1 Pulse Length and Amplitude	88
5.3.2 Pulse Speed	89
5.4 Graphs of Position and Velocity	90
5.4.1 Motion of a Particle of the Medium	90
5.4.2 Motion of the Pulse	92
5.5 Transmission and Reflection of a Pulse at a Boundary	96
5.6 Reflection of a Pulse from Fixed and Free Ends	97
5.6.1 Reflection of a Pulse from a Fixed End	97

5.6.2	Reflection of a Pulse from a Free End	98
5.7	Superposition of Pulses	99
5.8	Exercises - Transverse Pulses	102
6	Transverse Waves - Grade 10	105
6.1	Introduction	105
6.2	What is a <i>transverse wave</i> ?	105
6.2.1	Peaks and Troughs	106
6.2.2	Amplitude and Wavelength	107
6.2.3	Points in Phase	109
6.2.4	Period and Frequency	110
6.2.5	Speed of a Transverse Wave	111
6.3	Graphs of Particle Motion	115
6.4	Standing Waves and Boundary Conditions	118
6.4.1	Reflection of a Transverse Wave from a Fixed End	118
6.4.2	Reflection of a Transverse Wave from a Free End	118
6.4.3	Standing Waves	118
6.4.4	Nodes and anti-nodes	122
6.4.5	Wavelengths of Standing Waves with Fixed and Free Ends	122
6.4.6	Superposition and Interference	125
6.5	Summary	127
6.6	Exercises	127
7	Geometrical Optics - Grade 10	129
7.1	Introduction	129
7.2	Light Rays	129
7.2.1	Shadows	132
7.2.2	Ray Diagrams	132
7.3	Reflection	132
7.3.1	Terminology	133
7.3.2	Law of Reflection	133
7.3.3	Types of Reflection	135
7.4	Refraction	137
7.4.1	Refractive Index	139
7.4.2	Snell's Law	139
7.4.3	Apparent Depth	143
7.5	Mirrors	146
7.5.1	Image Formation	146
7.5.2	Plane Mirrors	147
7.5.3	Ray Diagrams	148
7.5.4	Spherical Mirrors	150
7.5.5	Concave Mirrors	150

7.5.6 Convex Mirrors	153
7.5.7 Summary of Properties of Mirrors	154
7.5.8 Magnification	154
7.6 Total Internal Reflection and Fibre Optics	156
7.6.1 Total Internal Reflection	156
7.6.2 Fibre Optics	161
7.7 Summary	163
7.8 Exercises	164
8 Magnetism - Grade 10	167
8.1 Introduction	167
8.2 Magnetic fields	167
8.3 Permanent magnets	169
8.3.1 The poles of permanent magnets	169
8.3.2 Magnetic attraction and repulsion	169
8.3.3 Representing magnetic fields	170
8.4 The compass and the earth's magnetic field	173
8.4.1 The earth's magnetic field	175
8.5 Summary	175
8.6 End of chapter exercises	176
9 Electrostatics - Grade 10	177
9.1 Introduction	177
9.2 Two kinds of charge	177
9.3 Unit of charge	177
9.4 Conservation of charge	177
9.5 Force between Charges	178
9.6 Conductors and insulators	181
9.6.1 The electroscope	182
9.7 Attraction between charged and uncharged objects	183
9.7.1 Polarisation of Insulators	183
9.8 Summary	184
9.9 End of chapter exercise	184
10 Electric Circuits - Grade 10	187
10.1 Electric Circuits	187
10.1.1 Closed circuits	187
10.1.2 Representing electric circuits	188
10.2 Potential Difference	192
10.2.1 Potential Difference	192
10.2.2 Potential Difference and Parallel Resistors	193
10.2.3 Potential Difference and Series Resistors	194
10.2.4 Ohm's Law	194

10.2.5 EMF	195
10.3 Current	198
10.3.1 Flow of Charge	198
10.3.2 Current	198
10.3.3 Series Circuits	199
10.3.4 Parallel Circuits	200
10.4 Resistance	202
10.4.1 What causes resistance?	202
10.4.2 Resistors in electric circuits	202
10.5 Instruments to Measure voltage, current and resistance	204
10.5.1 Voltmeter	204
10.5.2 Ammeter	204
10.5.3 Ohmmeter	204
10.5.4 Meters Impact on Circuit	205
10.6 Exercises - Electric circuits	205

III Grade 11 - Physics 209

11 Vectors	211
11.1 Introduction	211
11.2 Scalars and Vectors	211
11.3 Notation	211
11.3.1 Mathematical Representation	212
11.3.2 Graphical Representation	212
11.4 Directions	212
11.4.1 Relative Directions	212
11.4.2 Compass Directions	213
11.4.3 Bearing	213
11.5 Drawing Vectors	214
11.6 Mathematical Properties of Vectors	215
11.6.1 Adding Vectors	215
11.6.2 Subtracting Vectors	217
11.6.3 Scalar Multiplication	218
11.7 Techniques of Vector Addition	218
11.7.1 Graphical Techniques	218
11.7.2 Algebraic Addition and Subtraction of Vectors	223
11.8 Components of Vectors	228
11.8.1 Vector addition using components	231
11.8.2 Summary	235
11.8.3 End of chapter exercises: Vectors	236
11.8.4 End of chapter exercises: Vectors - Long questions	237

12 Force, Momentum and Impulse - Grade 11	239
12.1 Introduction	239
12.2 Force	239
12.2.1 What is a <i>force</i> ?	239
12.2.2 Examples of Forces in Physics	240
12.2.3 Systems and External Forces	241
12.2.4 Force Diagrams	242
12.2.5 Free Body Diagrams	243
12.2.6 Finding the Resultant Force	244
12.2.7 Exercise	246
12.3 Newton's Laws	246
12.3.1 Newton's First Law	247
12.3.2 Newton's Second Law of Motion	249
12.3.3 Exercise	261
12.3.4 Newton's Third Law of Motion	263
12.3.5 Exercise	267
12.3.6 Different types of forces	268
12.3.7 Exercise	275
12.3.8 Forces in equilibrium	276
12.3.9 Exercise	279
12.4 Forces between Masses	282
12.4.1 Newton's Law of Universal Gravitation	282
12.4.2 Comparative Problems	284
12.4.3 Exercise	286
12.5 Momentum and Impulse	287
12.5.1 Vector Nature of Momentum	290
12.5.2 Exercise	291
12.5.3 Change in Momentum	291
12.5.4 Exercise	293
12.5.5 Newton's Second Law revisited	293
12.5.6 Impulse	294
12.5.7 Exercise	296
12.5.8 Conservation of Momentum	297
12.5.9 Physics in Action: Impulse	300
12.5.10 Exercise	301
12.6 Torque and Levers	302
12.6.1 Torque	302
12.6.2 Mechanical Advantage and Levers	305
12.6.3 Classes of levers	307
12.6.4 Exercise	308
12.7 Summary	309
12.8 End of Chapter exercises	310

13 Geometrical Optics - Grade 11	327
13.1 Introduction	327
13.2 Lenses	327
13.2.1 Converging Lenses	329
13.2.2 Diverging Lenses	340
13.2.3 Summary of Image Properties	343
13.3 The Human Eye	344
13.3.1 Structure of the Eye	345
13.3.2 Defects of Vision	346
13.4 Gravitational Lenses	347
13.5 Telescopes	347
13.5.1 Refracting Telescopes	347
13.5.2 Reflecting Telescopes	348
13.5.3 Southern African Large Telescope	348
13.6 Microscopes	349
13.7 Summary	351
13.8 Exercises	352
 14 Longitudinal Waves - Grade 11	355
14.1 Introduction	355
14.2 What is a <i>longitudinal wave</i> ?	355
14.3 Characteristics of Longitudinal Waves	356
14.3.1 Compression and Rarefaction	356
14.3.2 Wavelength and Amplitude	357
14.3.3 Period and Frequency	357
14.3.4 Speed of a Longitudinal Wave	358
14.4 Graphs of Particle Position, Displacement, Velocity and Acceleration	359
14.5 Sound Waves	360
14.6 Seismic Waves	361
14.7 Summary - Longitudinal Waves	361
14.8 Exercises - Longitudinal Waves	362
 15 Sound - Grade 11	363
15.1 Introduction	363
15.2 Characteristics of a Sound Wave	363
15.2.1 Pitch	364
15.2.2 Loudness	364
15.2.3 Tone	364
15.3 Speed of Sound	365
15.4 Physics of the Ear and Hearing	365
15.4.1 Intensity of Sound	366
15.5 Ultrasound	367

15.6 SONAR	368
15.6.1 Echolocation	368
15.7 Summary	369
15.8 Exercises	369
16 The Physics of Music - Grade 11	373
16.1 Introduction	373
16.2 Standing Waves in String Instruments	373
16.3 Standing Waves in Wind Instruments	377
16.4 Resonance	382
16.5 Music and Sound Quality	384
16.6 Summary - The Physics of Music	385
16.7 End of Chapter Exercises	386
17 Electrostatics - Grade 11	387
17.1 Introduction	387
17.2 Forces between charges - Coulomb's Law	387
17.3 Electric field around charges	392
17.3.1 Electric field lines	393
17.3.2 Positive charge acting on a test charge	393
17.3.3 Combined charge distributions	394
17.3.4 Parallel plates	397
17.4 Electrical potential energy and potential	400
17.4.1 Electrical potential	400
17.4.2 Real-world application: lightning	402
17.5 Capacitance and the parallel plate capacitor	403
17.5.1 Capacitors and capacitance	403
17.5.2 Dielectrics	404
17.5.3 Physical properties of the capacitor and capacitance	404
17.5.4 Electric field in a capacitor	405
17.6 Capacitor as a circuit device	406
17.6.1 A capacitor in a circuit	406
17.6.2 Real-world applications: capacitors	407
17.7 Summary	407
17.8 Exercises - Electrostatics	407
18 Electromagnetism - Grade 11	413
18.1 Introduction	413
18.2 Magnetic field associated with a current	413
18.2.1 Real-world applications	418
18.3 Current induced by a changing magnetic field	420
18.3.1 Real-life applications	422
18.4 Transformers	423

18.4.1 Real-world applications	425
18.5 Motion of a charged particle in a magnetic field	425
18.5.1 Real-world applications	426
18.6 Summary	427
18.7 End of chapter exercises	427
19 Electric Circuits - Grade 11	429
19.1 Introduction	429
19.2 Ohm's Law	429
19.2.1 Definition of Ohm's Law	429
19.2.2 Ohmic and non-ohmic conductors	431
19.2.3 Using Ohm's Law	432
19.3 Resistance	433
19.3.1 Equivalent resistance	433
19.3.2 Use of Ohm's Law in series and parallel Circuits	438
19.3.3 Batteries and internal resistance	440
19.4 Series and parallel networks of resistors	442
19.5 Wheatstone bridge	445
19.6 Summary	447
19.7 End of chapter exercise	447
20 Electronic Properties of Matter - Grade 11	451
20.1 Introduction	451
20.2 Conduction	451
20.2.1 Metals	453
20.2.2 Insulator	453
20.2.3 Semi-conductors	454
20.3 Intrinsic Properties and Doping	454
20.3.1 Surplus	455
20.3.2 Deficiency	455
20.4 The p-n junction	457
20.4.1 Differences between p- and n-type semi-conductors	457
20.4.2 The p-n Junction	457
20.4.3 Unbiased	457
20.4.4 Forward biased	457
20.4.5 Reverse biased	458
20.4.6 Real-World Applications of Semiconductors	458
20.5 End of Chapter Exercises	459
IV Grade 12 - Physics	461
21 Motion in Two Dimensions - Grade 12	463
21.1 Introduction	463

21.2 Vertical Projectile Motion	463
21.2.1 Motion in a Gravitational Field	463
21.2.2 Equations of Motion	464
21.2.3 Graphs of Vertical Projectile Motion	467
21.3 Conservation of Momentum in Two Dimensions	475
21.4 Types of Collisions	480
21.4.1 Elastic Collisions	480
21.4.2 Inelastic Collisions	485
21.5 Frames of Reference	490
21.5.1 Introduction	490
21.5.2 What is a <i>frame of reference</i> ?	491
21.5.3 Why are frames of reference important?	491
21.5.4 Relative Velocity	491
21.6 Summary	494
21.7 End of chapter exercises	495
 22 Mechanical Properties of Matter - Grade 12	 503
22.1 Introduction	503
22.2 Deformation of materials	503
22.2.1 Hooke's Law	503
22.2.2 Deviation from Hooke's Law	506
22.3 Elasticity, plasticity, fracture, creep	508
22.3.1 Elasticity and plasticity	508
22.3.2 Fracture, creep and fatigue	508
22.4 Failure and strength of materials	509
22.4.1 The properties of matter	509
22.4.2 Structure and failure of materials	509
22.4.3 Controlling the properties of materials	509
22.4.4 Steps of Roman Swordsmithing	510
22.5 Summary	511
22.6 End of chapter exercise	511
 23 Work, Energy and Power - Grade 12	 513
23.1 Introduction	513
23.2 Work	513
23.3 Energy	519
23.3.1 External and Internal Forces	519
23.3.2 Capacity to do Work	520
23.4 Power	525
23.5 Important Equations and Quantities	529
23.6 End of Chapter Exercises	529

24 Doppler Effect - Grade 12	533
24.1 Introduction	533
24.2 The Doppler Effect with Sound and Ultrasound	533
24.2.1 Ultrasound and the Doppler Effect	537
24.3 The Doppler Effect with Light	537
24.3.1 The Expanding Universe	538
24.4 Summary	539
24.5 End of Chapter Exercises	539
 25 Colour - Grade 12	541
25.1 Introduction	541
25.2 Colour and Light	541
25.2.1 Dispersion of white light	544
25.3 Addition and Subtraction of Light	544
25.3.1 Additive Primary Colours	544
25.3.2 Subtractive Primary Colours	545
25.3.3 Complementary Colours	546
25.3.4 Perception of Colour	546
25.3.5 Colours on a Television Screen	547
25.4 Pigments and Paints	548
25.4.1 Colour of opaque objects	548
25.4.2 Colour of transparent objects	548
25.4.3 Pigment primary colours	549
25.5 End of Chapter Exercises	550
 26 2D and 3D Wavefronts - Grade 12	553
26.1 Introduction	553
26.2 Wavefronts	553
26.3 The Huygens Principle	554
26.4 Interference	556
26.5 Diffraction	557
26.5.1 Diffraction through a Slit	558
26.6 Shock Waves and Sonic Booms	562
26.6.1 Subsonic Flight	563
26.6.2 Supersonic Flight	563
26.6.3 Mach Cone	566
26.7 End of Chapter Exercises	568
 27 Wave Nature of Matter - Grade 12	571
27.1 Introduction	571
27.2 de Broglie Wavelength	571
27.3 The Electron Microscope	574
27.3.1 Disadvantages of an Electron Microscope	577

27.3.2 Uses of Electron Microscopes	577
27.4 End of Chapter Exercises	578
28 Electrodynamics - Grade 12	579
28.1 Introduction	579
28.2 Electrical machines - generators and motors	579
28.2.1 Electrical generators	580
28.2.2 Electric motors	582
28.2.3 Real-life applications	582
28.2.4 Exercise - generators and motors	584
28.3 Alternating Current	585
28.3.1 Exercise - alternating current	586
28.4 Capacitance and inductance	586
28.4.1 Capacitance	586
28.4.2 Inductance	586
28.4.3 Exercise - capacitance and inductance	588
28.5 Summary	588
28.6 End of chapter exercise	589
29 Electronics - Grade 12	591
29.1 Introduction	591
29.2 Capacitive and Inductive Circuits	591
29.3 Filters and Signal Tuning	596
29.3.1 Capacitors and Inductors as Filters	596
29.3.2 LRC Circuits, Resonance and Signal Tuning	596
29.4 Active Circuit Elements	599
29.4.1 The Diode	599
29.4.2 The Light Emitting Diode (LED)	601
29.4.3 Transistor	603
29.4.4 The Operational Amplifier	607
29.5 The Principles of Digital Electronics	609
29.5.1 Logic Gates	610
29.6 Using and Storing Binary Numbers	616
29.6.1 Binary numbers	616
29.6.2 Counting circuits	617
29.6.3 Storing binary numbers	619
30 EM Radiation	625
30.1 Introduction	625
30.2 Particle/wave nature of electromagnetic radiation	625
30.3 The wave nature of electromagnetic radiation	626
30.4 Electromagnetic spectrum	626
30.5 The particle nature of electromagnetic radiation	629

30.5.1 Exercise - particle nature of EM waves	630
30.6 Penetrating ability of electromagnetic radiation	631
30.6.1 Ultraviolet(UV) radiation and the skin	631
30.6.2 Ultraviolet radiation and the eyes	632
30.6.3 X-rays	632
30.6.4 Gamma-rays	632
30.6.5 Exercise - Penetrating ability of EM radiation	633
30.7 Summary	633
30.8 End of chapter exercise	633
31 Optical Phenomena and Properties of Matter - Grade 12	635
31.1 Introduction	635
31.2 The transmission and scattering of light	635
31.2.1 Energy levels of an electron	635
31.2.2 Interaction of light with metals	636
31.2.3 Why is the sky blue?	637
31.3 The photoelectric effect	638
31.3.1 Applications of the photoelectric effect	640
31.3.2 Real-life applications	642
31.4 Emission and absorption spectra	643
31.4.1 Emission Spectra	643
31.4.2 Absorption spectra	644
31.4.3 Colours and energies of electromagnetic radiation	646
31.4.4 Applications of emission and absorption spectra	648
31.5 Lasers	650
31.5.1 How a laser works	652
31.5.2 A simple laser	654
31.5.3 Laser applications and safety	655
31.6 Summary	656
31.7 End of chapter exercise	657
V Exercises	659
32 Exercises	661
VI Essays	663
Essay 1: Energy and electricity. Why the fuss?	665
33 Essay: How a cell phone works	671
34 Essay: How a Physiotherapist uses the Concept of Levers	673
35 Essay: How a Pilot Uses Vectors	675

Chapter 25

Colour - Grade 12

25.1 Introduction

We call the light that we humans can see 'visible light'. Visible light is actually just a small part of the large spectrum of electromagnetic radiation which you will learn more about in Chapter 30. We can think of electromagnetic radiation and visible light as transverse waves. We know that transverse waves can be described by their amplitude, frequency (or wavelength) and velocity. The velocity of a wave is given by the product of its frequency and wavelength:

$$v = f \times \lambda \quad (25.1)$$

However, electromagnetic radiation, including visible light, is special because, no matter what the frequency, it all moves at a **constant velocity** (in vacuum) which is known as the speed of light. The speed of light has the symbol c and is:

$$c = 3 \times 10^8 \text{ m.s}^{-1}$$

Since the *speed of light* is c , we can then say:

$$c = f \times \lambda \quad (25.2)$$

25.2 Colour and Light

Our eyes are sensitive to visible light over a range of wavelengths from 390 nm to 780 nm ($1 \text{ nm} = 1 \times 10^{-9} \text{ m}$). The different **colours** of light we see are related to specific *frequencies* (and *wavelengths*) of visible light. The wavelengths and frequencies are listed in table 25.1.

Colour	Wavelength range (nm)	Frequency range (Hz)
violet	390 - 455	$769 - 659 \times 10^{12}$
blue	455 - 492	$659 - 610 \times 10^{12}$
green	492 - 577	$610 - 520 \times 10^{12}$
yellow	577 - 597	$520 - 503 \times 10^{12}$
orange	597 - 622	$503 - 482 \times 10^{12}$
red	622 - 780	$482 - 385 \times 10^{12}$

Table 25.1: Colours, wavelengths and frequencies of light in the visible spectrum.

You can see from table 25.1 that **violet** light has the *shortest wavelengths* and *highest frequencies* while **red** light has the *longest wavelengths* and *lowest frequencies*.



Worked Example 159: Calculating the frequency of light given the wavelength

Question: A streetlight emits light with a wavelength of 520 nm.

1. What colour is the light? (Use table 25.1 to determine the colour)
2. What is the frequency of the light?

Answer

Step 1 : What is being asked and what information are we given?

We need to determine the colour and frequency of light with a wavelength of $\lambda = 520 \text{ nm} = 520 \times 10^{-9} \text{ m}$.

Step 2 : Compare the wavelength of the light to those given in table 25.1

We see from table 25.1 that light with wavelengths between 492 - 577 nm is green. 520 nm falls into this range, therefore the colour of the light is green.

Step 3 : Next we need to calculate the frequency of the light

We know that

$$c = f \times \lambda$$

We know c and we are given that $\lambda = 520 \times 10^{-9} \text{ m}$. So we can substitute in these values and solve for the frequency f . (**NOTE:** Don't forget to always change units into S.I. units! $1 \text{ nm} = 1 \times 10^{-9} \text{ m}$.)

$$\begin{aligned} f &= \frac{c}{\lambda} \\ &= \frac{3 \times 10^8}{520 \times 10^{-9}} \\ &= 577 \times 10^{12} \text{ Hz} \end{aligned}$$

The frequency of the green light is $577 \times 10^{12} \text{ Hz}$



Worked Example 160: Calculating the wavelength of light given the frequency

Question: A streetlight also emits light with a frequency of $490 \times 10^{12} \text{ Hz}$.

1. What colour is the light? (Use table 25.1 to determine the colour)
2. What is the wavelength of the light?

Answer

Step 1 : What is being asked and what information are we given?

We need to find the colour and wavelength of light which has a frequency of $490 \times 10^{12} \text{ Hz}$ and which is emitted by the streetlight.

Step 2 : Compare the wavelength of the light to those given in table 25.1

We can see from table 25.1 that orange light has frequencies between $503 - 482 \times 10^{12} \text{ Hz}$. The light from the streetlight has $f = 490 \times 10^{12} \text{ Hz}$ which fits into this range. Therefore the light must be orange in colour.

Step 3 : Next we need to calculate the wavelength of the light

We know that

$$c = f \times \lambda$$

We know $c = 3 \times 10^8 \text{ m.s}^{-1}$ and we are given that $f = 490 \times 10^{12} \text{ Hz}$. So we can

substitute in these values and solve for the wavelength λ .

$$\begin{aligned}\lambda &= \frac{c}{f} \\ &= \frac{3 \times 10^8}{490 \times 10^{12}} \\ &= 6.122 \times 10^{-7} \text{ m} \\ &= 612 \times 10^{-9} \text{ m} \\ &= 612 \text{ nm}\end{aligned}$$

Therefore the orange light has a wavelength of 612 nm.



Worked Example 161: Frequency of Green

Question: The wavelength of green light ranges between 500 nm and 565 nm. Calculate the range of frequencies that correspond to this range of wavelengths.

Answer

Step 1 : Determine how to approach the problem

Use

$$c = f \times \lambda$$

to determine f .

Step 2 : Calculate frequency corresponding to upper limit of wavelength range

$$\begin{aligned}c &= f \times \lambda \\ f &= \frac{c}{\lambda} \\ &= \frac{3 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{565 \times 10^{-9} \text{ m}} \\ &= 5,31 \times 10^{14} \text{ Hz}\end{aligned}$$

Step 3 : Calculate frequency corresponding to lower limit of wavelength range

$$\begin{aligned}c &= f \times \lambda \\ f &= \frac{c}{\lambda} \\ &= \frac{3 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{500 \times 10^{-9} \text{ m}} \\ &= 6,00 \times 10^{14} \text{ Hz}\end{aligned}$$

Step 4 : Write final answer

The range of frequencies of green light is $5,31 \times 10^{14}$ Hz to $6,00 \times 10^{14}$ Hz.



Exercise: Calculating wavelengths and frequencies of light

- Calculate the frequency of light which has a wavelength of 400 nm.
(Remember to use S.I. units)

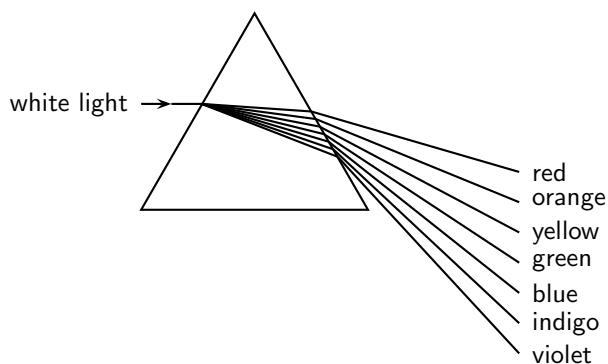
2. Calculate the wavelength of light which has a frequency of 550×10^{12} Hz.
 3. What colour is light which has a wavelength of 470×10^9 m and what is its frequency?
 4. What is the wavelength of light with a frequency of 510×10^{12} Hz and what is its color?
-

25.2.1 Dispersion of white light

White light, like the light which comes from the sun, is made up of *all* the visible wavelengths of light. In other words, white light is a *combination* of all the colours of visible light.

In Chapter 7, you learnt that the speed of light is different in different substances. The speed of light in different substances depends on the frequency of the light. For example, when white light travels through glass, light of the different frequencies is slowed down by different amounts. The lower the frequency, the less the speed is reduced which means that red light (lowest frequency) is slowed down *less* than violet light (highest frequency). We can see this when white light is incident on a glass prism.

Have a look at the picture below. When the white light hits the edge of the prism, the light which travels through the glass is refracted as it moves from the less dense medium (air) to the more dense medium (glass).



- The **red** light which is slowed down the *least*, is refracted the *least*.
- The **violet** light which is slowed down the *most*, is refracted the *most*.

When the light hits the other side of the prism it is again refracted but the angle of the prism edge allows the light to remain separated into its different colours. White light is therefore separated into its different colours by the prism and we say that the white light has been **dispersed** by the prism.

The dispersion effect is also responsible for why we see rainbows. When sunlight hits drops of water in the atmosphere, the white light is dispersed into its different colours by the water.

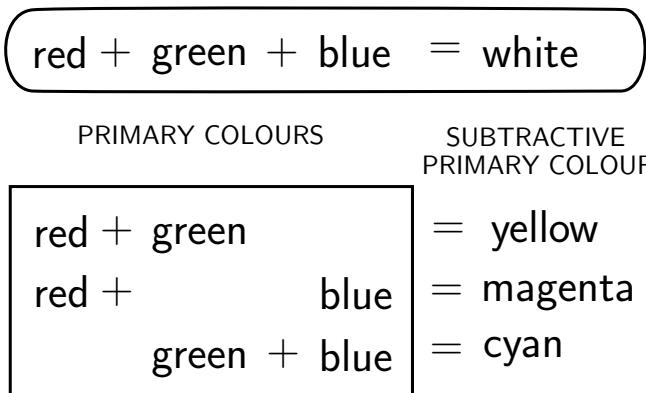
25.3 Addition and Subtraction of Light

25.3.1 Additive Primary Colours

The primary colours of light are **red**, **green** and **blue**. When all the primary colours are superposed (added together), white light is produced. Red, green and blue are therefore called the *additive primary colours*. All the other colours can be produced by different combinations of red, green and blue.

25.3.2 Subtractive Primary Colours

The subtractive primary colours are obtained by subtracting one of the three additive primary colours from white light. The subtractive primary colours are **yellow**, **magenta** and **cyan**. Magenta appears as a pinkish-purplish colour and cyan looks greenish-blue. You can see how the primary colours of light add up to the different subtractive colours in the illustration below.



Activity :: Experiment : Colours of light

Aim:

To investigate the additive properties of colours and determine the complementary colours of light.

Apparatus:

You will need two battery operated torches with flat bulb fronts, a large piece of white paper, and some pieces of cellophane paper of the following colours: red, blue, green, yellow, cyan, magenta. (You should easily be able to get these from a newsagents.)

Make a table in your workbook like the one below:

Colour 1	Colour 2	Final colour prediction	Final colour measured
red	blue		
red	green		
green	blue		
magenta	green		
yellow	blue		
cyan	red		

Before you begin your experiment, use what you know about colours of light to write down in the third column "Final colour prediction", what you think the result of adding the two colours of light will be. You will then be able to test your predictions by making the following measurements:

Method:

Proceed according to the table above. Put the correct colour of cellophane paper over each torch bulb. e.g. the first test will be to put red cellophane on one torch and blue cellophane on the other. Switch on the torch with the red cellophane over it and shine it onto the piece of white paper.

What colour is the light?

Turn off that torch and turn on the one with blue cellophane and shine it onto the white paper.

What colour is the light?

Now shine both torches with their cellophane coverings onto the same spot on the white paper. What is the colour of the light produced? Write this down in the fourth column of your table.

Repeat the experiment for the other colours of cellophane so that you can complete your table.

Questions:

1. How did your predictions match up to your measurements?
 2. Complementary colours of light are defined as the colours of light which, when added to one of the primary colours, produce white light. From your completed table, write down the complementary colours for red, blue and green.
-

25.3.3 Complementary Colours

Complementary colours are two colours of light which add together to give white.

Activity :: Investigation : Complementary colours for red, green and blue

Complementary colours are two colours which add together to give white. Place a tick in the box where the colours in the first column added to the colours in the top row give white.

	magenta (=red+blue)	yellow (=red+green)	cyan (=blue+green)
red			
green			
blue			

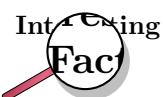
You should have found that the complementary colours for red, green and blue are:

- Red and Cyan
- Green and Magenta
- Blue and Yellow

25.3.4 Perception of Colour

The light-sensitive lining on the back inside half of the human eye is called the retina. The retina contains two kinds of light sensitive cells or *photoreceptors*: the rod cells (sensitive to low light) and the cone cells (sensitive to normal daylight) which enable us to see. The rods are not very sensitive to colour but work well in dimly lit conditions. This is why it is possible to see in a dark room, but it is hard to see any colours. Only your rods are sensitive to the low light levels and so you can only see in black, white and grey. The cones enable us to see colours. Normally, there are three kinds of cones, each containing a different pigment. The cones are activated when the pigments absorb light. The three types of cones are sensitive to (i.e. absorb) red, blue and green light respectively. Therefore we can perceive *all* the different colours in the visible spectrum when the different types of cones are stimulated by different amounts since they are just combinations of the three primary colours of light.

The rods and cones have different response times to light. The cones react quickly when bright light falls on them. The rods take a longer time to react. This is why it takes a while (about 10 minutes) for your eyes to adjust when you enter a dark room after being outside on a sunny day.



Color blindness in humans is the inability to perceive differences between some or all colors that other people can see. Most often it is a genetic problem, but may also occur because of eye, nerve, or brain damage, or due to exposure to certain chemicals. The most common forms of human color blindness result from problems with either the middle or long wavelength sensitive cone systems, and involve difficulties in discriminating reds, yellows, and greens from one another. This is called "red-green color blindness". Other forms of color blindness are much rarer. They include problems in discriminating blues from yellows, and the rarest forms of all, complete color blindness or monochromasy, where one cannot distinguish any color from grey, as in a black-and-white movie or photograph.



Worked Example 162: Seeing Colours

Question: When blue and green light fall on an eye, is cyan light being created?
Discuss.

Answer

Cyan light is not created when blue and green light fall on the eye. The blue and green receptors are stimulated to make the brain believe that cyan light is being created.

25.3.5 Colours on a Television Screen

If you look very closely at a colour cathode-ray television screen or computer screen, you will see that there are very many small red, green and blue dots called *phosphors* on it. These dots are caused to fluoresce (glow brightly) when a beam of electrons from the cathode-ray tube behind the screen hits them. Since different combinations of the three primary colours of light can produce any other colour, only red, green and blue dots are needed to make pictures containing all the colours of the visible spectrum.



Exercise: Colours of light

1. List the three primary colours of light.
2. What is the term for the phenomenon whereby white light is split up into its different colours by a prism?
3. What is meant by the term "complementary colour" of light?
4. When white light strikes a prism which colour of light is refracted the most and which is refracted the least? Explain your answer in terms of the speed of light in a medium.

25.4 Pigments and Paints

We have learnt that white light is a combination of all the colours of the visible spectrum and that each colour of light is related to a different frequency. But what gives everyday objects around us their different colours?

Pigments are substances which give an object its colour by absorbing certain frequencies of light and reflecting other frequencies. For example, a red pigment absorbs all colours of light except red which it reflects. Paints and inks contain pigments which gives the paints and inks different colours.

25.4.1 Colour of opaque objects

Objects which you *cannot* see through (i.e. they are not transparent) are called **opaque**.

Examples of some opaque objects are metals, wood and bricks. The colour of an opaque object is determined by the colours (therefore *frequencies*) of light which it *reflects*. For example, when white light strikes a blue opaque object such as a ruler, the ruler will absorb all frequencies of light *except* blue, which will be reflected. The reflected blue light is the light which makes it into our eyes and therefore the object will appear blue.

Opaque objects which appear white do not absorb any light. They reflect all the frequencies. Black opaque objects absorb all frequencies of light. They do not reflect at all and therefore appear to have no colour.

Worked Example 163: Colour of Opaque Objects

Question: If we shine white light on a sheet of paper that can only reflect green light, what is the colour of the paper?

Answer

Since the colour of an object is determined by that frequency of light that is *reflected*, the sheet of paper will appear green, as this is the only frequency that is reflected. All the other frequencies are absorbed by the paper.

Worked Example 164: Colour of an opaque object II

Question: The cover of a book appears to have a magenta colour. What colours of light does it reflect and what colours does it absorb?

Answer

We know that magenta is a combination of red and blue primary colours of light. Therefore the object must be reflecting blue and red light and absorb green.

25.4.2 Colour of transparent objects

If an object is **transparent** it means that you can see through it. For example, glass, clean water and some clear plastics are transparent. The colour of a transparent object is determined by the colours (frequencies) of light which it *transmits* (allows to pass through it). For example, a cup made of green glass will appear green because it absorbs all the other frequencies of light *except* green, which it transmits. This is the light which we receive in our eyes and the object appears green.



Worked Example 165: Colour of Transparent Objects

Question: If white light is shone through a glass plate that absorbs light of all frequencies except red, what is the colour of the glass plate?

Answer

Since the colour of an object is determined by that frequency of light that is *transmitted*, the glass plate will appear red, as this is the only frequency that is not absorbed.

25.4.3 Pigment primary colours

The primary pigments and paints are **cyan**, **magenta** and **yellow**. When pigments or paints of these three colours are mixed together in equal amounts they produce **black**. Any other colour of paint can be made by mixing the primary pigments together in different quantities. The primary pigments are related to the primary colours of light in the following way:

PRIMARY PIGMENTS		PRIMARY COLOURS OF LIGHT
cyan	+ magenta + yellow = black	
cyan + magenta	= blue	
cyan + yellow	= green	
magenta + yellow	= red	



Colour printers only use 4 colours of ink: cyan, magenta, yellow and black. All the other colours can be mixed from these!



Worked Example 166: Pigments

Question: What colours of light are absorbed by a green pigment?

Answer

If the pigment is green, then green light must be *reflected*. Therefore, red and blue light are absorbed.



Worked Example 167: Primary pigments

Question: I have a ruler which reflects red light and absorbs all other colours of light. What colour does the ruler appear in white light? What primary pigments must have been mixed to make the pigment which gives the ruler its colour?

Answer

Step 1 : What is being asked and what are we given?

We need to determine the colour of the ruler and the pigments which were mixed to make the colour.

Step 2 : An opaque object appears the colour of the light it reflects

The ruler reflects red light and absorbs all other colours. Therefore the ruler appears to be red.

Step 3 : What pigments need to be mixed to get red?

Red pigment is produced when magenta and yellow pigments are mixed. Therefore magenta and yellow pigments were mixed to make the red pigment which gives the ruler its colour.



Worked Example 168: Paint Colours

Question: If cyan light shines on a dress that contains a pigment that is capable of absorbing blue, what colour does the dress appear?

Answer

Step 1 : Determine the component colours of cyan light

Cyan light is made up of blue and green light.

Step 2 : Determine solution

If the dress absorbs the blue light then the green light must be reflected, so the dress will appear green!

25.5 End of Chapter Exercises

1. Calculate the wavelength of light which has a frequency of 570×10^{12} Hz.
2. Calculate the frequency of light which has a wavelength of 580 nm.
3. Complete the following sentence: When white light is dispersed by a prism, light of the colour ? is refracted the most and light of colour ? is refracted the least.
4. What are the two types of photoreceptor found in the retina of the human eye called and which type is sensitive to colours?
5. What color do the following shirts appear to the human eye when the lights in a room are turned off and the room is completely dark?
 - A red shirt
 - B blue shirt
 - C green shirt
6. Two light bulbs, each of a different colour, shine on a sheet of white paper. Each light bulb can be a primary colour of light - red, green, and blue. Depending on which primary colour of light is used, the paper will appear a different color. What colour will the paper appear if the lights are:
 - A red and blue?
 - B red and green?

C green and blue?

7. Match the primary colour of light on the left to its complementary colour on the right:

Column A	Column B
red	yellow
green	cyan
blue	magenta

8. Which combination of colours of light gives magenta?

A red and yellow

B green and red

C blue and cyan

D blue and red

9. Which combination of colours of light gives cyan?

A yellow and red

B green and blue

C blue and magenta

D blue and red

10. If yellow light falls on an object whose pigment absorbs green light, what colour will the object appear?

11. If yellow light falls on a blue pigment, what colour will it appear?

Appendix A

GNU Free Documentation License

Version 1.2, November 2002

Copyright © 2000,2001,2002 Free Software Foundation, Inc.

59 Temple Place, Suite 330, Boston, MA 02111-1307 USA

Everyone is permitted to copy and distribute verbatim copies of this license document, but changing it is not allowed.

PREAMBLE

The purpose of this License is to make a manual, textbook, or other functional and useful document "free" in the sense of freedom: to assure everyone the effective freedom to copy and redistribute it, with or without modifying it, either commercially or non-commercially. Secondary, this License preserves for the author and publisher a way to get credit for their work, while not being considered responsible for modifications made by others.

This License is a kind of "copyleft", which means that derivative works of the document must themselves be free in the same sense. It complements the GNU General Public License, which is a copyleft license designed for free software.

We have designed this License in order to use it for manuals for free software, because free software needs free documentation: a free program should come with manuals providing the same freedoms that the software does. But this License is not limited to software manuals; it can be used for any textual work, regardless of subject matter or whether it is published as a printed book. We recommend this License principally for works whose purpose is instruction or reference.

APPLICABILITY AND DEFINITIONS

This License applies to any manual or other work, in any medium, that contains a notice placed by the copyright holder saying it can be distributed under the terms of this License. Such a notice grants a world-wide, royalty-free license, unlimited in duration, to use that work under the conditions stated herein. The "Document", below, refers to any such manual or work. Any member of the public is a licensee, and is addressed as "you". You accept the license if you copy, modify or distribute the work in a way requiring permission under copyright law.

A "Modified Version" of the Document means any work containing the Document or a portion of it, either copied verbatim, or with modifications and/or translated into another language.

A "Secondary Section" is a named appendix or a front-matter section of the Document that deals exclusively with the relationship of the publishers or authors of the Document to the Document's overall subject (or to related matters) and contains nothing that could fall directly within that overall subject. (Thus, if the Document is in part a textbook of mathematics, a

Secondary Section may not explain any mathematics.) The relationship could be a matter of historical connection with the subject or with related matters, or of legal, commercial, philosophical, ethical or political position regarding them.

The “Invariant Sections” are certain Secondary Sections whose titles are designated, as being those of Invariant Sections, in the notice that says that the Document is released under this License. If a section does not fit the above definition of Secondary then it is not allowed to be designated as Invariant. The Document may contain zero Invariant Sections. If the Document does not identify any Invariant Sections then there are none.

The “Cover Texts” are certain short passages of text that are listed, as Front-Cover Texts or Back-Cover Texts, in the notice that says that the Document is released under this License. A Front-Cover Text may be at most 5 words, and a Back-Cover Text may be at most 25 words.

A “Transparent” copy of the Document means a machine-readable copy, represented in a format whose specification is available to the general public, that is suitable for revising the document straightforwardly with generic text editors or (for images composed of pixels) generic paint programs or (for drawings) some widely available drawing editor, and that is suitable for input to text formatters or for automatic translation to a variety of formats suitable for input to text formatters. A copy made in an otherwise Transparent file format whose markup, or absence of markup, has been arranged to thwart or discourage subsequent modification by readers is not Transparent. An image format is not Transparent if used for any substantial amount of text. A copy that is not “Transparent” is called “Opaque”.

Examples of suitable formats for Transparent copies include plain ASCII without markup, Texinfo input format, L^AT_EX input format, SGML or XML using a publicly available DTD and standard-conforming simple HTML, PostScript or PDF designed for human modification. Examples of transparent image formats include PNG, XCF and JPG. Opaque formats include proprietary formats that can be read and edited only by proprietary word processors, SGML or XML for which the DTD and/or processing tools are not generally available, and the machine-generated HTML, PostScript or PDF produced by some word processors for output purposes only.

The “Title Page” means, for a printed book, the title page itself, plus such following pages as are needed to hold, legibly, the material this License requires to appear in the title page. For works in formats which do not have any title page as such, “Title Page” means the text near the most prominent appearance of the work’s title, preceding the beginning of the body of the text.

A section “Entitled XYZ” means a named subunit of the Document whose title either is precisely XYZ or contains XYZ in parentheses following text that translates XYZ in another language. (Here XYZ stands for a specific section name mentioned below, such as “Acknowledgements”, “Dedications”, “Endorsements”, or “History”.) To “Preserve the Title” of such a section when you modify the Document means that it remains a section “Entitled XYZ” according to this definition.

The Document may include Warranty Disclaimers next to the notice which states that this License applies to the Document. These Warranty Disclaimers are considered to be included by reference in this License, but only as regards disclaiming warranties: any other implication that these Warranty Disclaimers may have is void and has no effect on the meaning of this License.

VERBATIM COPYING

You may copy and distribute the Document in any medium, either commercially or non-commercially, provided that this License, the copyright notices, and the license notice saying this License applies to the Document are reproduced in all copies, and that you add no other conditions whatsoever to those of this License. You may not use technical measures to obstruct or control the reading or further copying of the copies you make or distribute. However, you may accept compensation in exchange for copies. If you distribute a large enough number of copies you must also follow the conditions in section A.

You may also lend copies, under the same conditions stated above, and you may publicly display copies.

COPYING IN QUANTITY

If you publish printed copies (or copies in media that commonly have printed covers) of the Document, numbering more than 100, and the Document's license notice requires Cover Texts, you must enclose the copies in covers that carry, clearly and legibly, all these Cover Texts: Front-Cover Texts on the front cover, and Back-Cover Texts on the back cover. Both covers must also clearly and legibly identify you as the publisher of these copies. The front cover must present the full title with all words of the title equally prominent and visible. You may add other material on the covers in addition. Copying with changes limited to the covers, as long as they preserve the title of the Document and satisfy these conditions, can be treated as verbatim copying in other respects.

If the required texts for either cover are too voluminous to fit legibly, you should put the first ones listed (as many as fit reasonably) on the actual cover, and continue the rest onto adjacent pages.

If you publish or distribute Opaque copies of the Document numbering more than 100, you must either include a machine-readable Transparent copy along with each Opaque copy, or state in or with each Opaque copy a computer-network location from which the general network-using public has access to download using public-standard network protocols a complete Transparent copy of the Document, free of added material. If you use the latter option, you must take reasonably prudent steps, when you begin distribution of Opaque copies in quantity, to ensure that this Transparent copy will remain thus accessible at the stated location until at least one year after the last time you distribute an Opaque copy (directly or through your agents or retailers) of that edition to the public.

It is requested, but not required, that you contact the authors of the Document well before redistributing any large number of copies, to give them a chance to provide you with an updated version of the Document.

MODIFICATIONS

You may copy and distribute a Modified Version of the Document under the conditions of sections A and A above, provided that you release the Modified Version under precisely this License, with the Modified Version filling the role of the Document, thus licensing distribution and modification of the Modified Version to whoever possesses a copy of it. In addition, you must do these things in the Modified Version:

1. Use in the Title Page (and on the covers, if any) a title distinct from that of the Document, and from those of previous versions (which should, if there were any, be listed in the History section of the Document). You may use the same title as a previous version if the original publisher of that version gives permission.
2. List on the Title Page, as authors, one or more persons or entities responsible for authorship of the modifications in the Modified Version, together with at least five of the principal authors of the Document (all of its principal authors, if it has fewer than five), unless they release you from this requirement.
3. State on the Title page the name of the publisher of the Modified Version, as the publisher.
4. Preserve all the copyright notices of the Document.
5. Add an appropriate copyright notice for your modifications adjacent to the other copyright notices.

6. Include, immediately after the copyright notices, a license notice giving the public permission to use the Modified Version under the terms of this License, in the form shown in the Addendum below.
7. Preserve in that license notice the full lists of Invariant Sections and required Cover Texts given in the Document's license notice.
8. Include an unaltered copy of this License.
9. Preserve the section Entitled "History", Preserve its Title, and add to it an item stating at least the title, year, new authors, and publisher of the Modified Version as given on the Title Page. If there is no section Entitled "History" in the Document, create one stating the title, year, authors, and publisher of the Document as given on its Title Page, then add an item describing the Modified Version as stated in the previous sentence.
10. Preserve the network location, if any, given in the Document for public access to a Transparent copy of the Document, and likewise the network locations given in the Document for previous versions it was based on. These may be placed in the "History" section. You may omit a network location for a work that was published at least four years before the Document itself, or if the original publisher of the version it refers to gives permission.
11. For any section Entitled "Acknowledgements" or "Dedications", Preserve the Title of the section, and preserve in the section all the substance and tone of each of the contributor acknowledgements and/or dedications given therein.
12. Preserve all the Invariant Sections of the Document, unaltered in their text and in their titles. Section numbers or the equivalent are not considered part of the section titles.
13. Delete any section Entitled "Endorsements". Such a section may not be included in the Modified Version.
14. Do not re-title any existing section to be Entitled "Endorsements" or to conflict in title with any Invariant Section.
15. Preserve any Warranty Disclaimers.

If the Modified Version includes new front-matter sections or appendices that qualify as Secondary Sections and contain no material copied from the Document, you may at your option designate some or all of these sections as invariant. To do this, add their titles to the list of Invariant Sections in the Modified Version's license notice. These titles must be distinct from any other section titles.

You may add a section Entitled "Endorsements", provided it contains nothing but endorsements of your Modified Version by various parties—for example, statements of peer review or that the text has been approved by an organisation as the authoritative definition of a standard.

You may add a passage of up to five words as a Front-Cover Text, and a passage of up to 25 words as a Back-Cover Text, to the end of the list of Cover Texts in the Modified Version. Only one passage of Front-Cover Text and one of Back-Cover Text may be added by (or through arrangements made by) any one entity. If the Document already includes a cover text for the same cover, previously added by you or by arrangement made by the same entity you are acting on behalf of, you may not add another; but you may replace the old one, on explicit permission from the previous publisher that added the old one.

The author(s) and publisher(s) of the Document do not by this License give permission to use their names for publicity for or to assert or imply endorsement of any Modified Version.

COMBINING DOCUMENTS

You may combine the Document with other documents released under this License, under the terms defined in section A above for modified versions, provided that you include in the

combination all of the Invariant Sections of all of the original documents, unmodified, and list them all as Invariant Sections of your combined work in its license notice, and that you preserve all their Warranty Disclaimers.

The combined work need only contain one copy of this License, and multiple identical Invariant Sections may be replaced with a single copy. If there are multiple Invariant Sections with the same name but different contents, make the title of each such section unique by adding at the end of it, in parentheses, the name of the original author or publisher of that section if known, or else a unique number. Make the same adjustment to the section titles in the list of Invariant Sections in the license notice of the combined work.

In the combination, you must combine any sections Entitled "History" in the various original documents, forming one section Entitled "History"; likewise combine any sections Entitled "Acknowledgements", and any sections Entitled "Dedications". You must delete all sections Entitled "Endorsements".

COLLECTIONS OF DOCUMENTS

You may make a collection consisting of the Document and other documents released under this License, and replace the individual copies of this License in the various documents with a single copy that is included in the collection, provided that you follow the rules of this License for verbatim copying of each of the documents in all other respects.

You may extract a single document from such a collection, and distribute it individually under this License, provided you insert a copy of this License into the extracted document, and follow this License in all other respects regarding verbatim copying of that document.

AGGREGATION WITH INDEPENDENT WORKS

A compilation of the Document or its derivatives with other separate and independent documents or works, in or on a volume of a storage or distribution medium, is called an "aggregate" if the copyright resulting from the compilation is not used to limit the legal rights of the compilation's users beyond what the individual works permit. When the Document is included in an aggregate, this License does not apply to the other works in the aggregate which are not themselves derivative works of the Document.

If the Cover Text requirement of section A is applicable to these copies of the Document, then if the Document is less than one half of the entire aggregate, the Document's Cover Texts may be placed on covers that bracket the Document within the aggregate, or the electronic equivalent of covers if the Document is in electronic form. Otherwise they must appear on printed covers that bracket the whole aggregate.

TRANSLATION

Translation is considered a kind of modification, so you may distribute translations of the Document under the terms of section A. Replacing Invariant Sections with translations requires special permission from their copyright holders, but you may include translations of some or all Invariant Sections in addition to the original versions of these Invariant Sections. You may include a translation of this License, and all the license notices in the Document, and any Warranty Disclaimers, provided that you also include the original English version of this License and the original versions of those notices and disclaimers. In case of a disagreement between the translation and the original version of this License or a notice or disclaimer, the original version will prevail.

If a section in the Document is Entitled "Acknowledgements", "Dedications", or "History", the requirement (section A) to Preserve its Title (section A) will typically require changing the

actual title.

TERMINATION

You may not copy, modify, sub-license, or distribute the Document except as expressly provided for under this License. Any other attempt to copy, modify, sub-license or distribute the Document is void, and will automatically terminate your rights under this License. However, parties who have received copies, or rights, from you under this License will not have their licenses terminated so long as such parties remain in full compliance.

FUTURE REVISIONS OF THIS LICENSE

The Free Software Foundation may publish new, revised versions of the GNU Free Documentation License from time to time. Such new versions will be similar in spirit to the present version, but may differ in detail to address new problems or concerns. See <http://www.gnu.org/copyleft/>.

Each version of the License is given a distinguishing version number. If the Document specifies that a particular numbered version of this License "or any later version" applies to it, you have the option of following the terms and conditions either of that specified version or of any later version that has been published (not as a draft) by the Free Software Foundation. If the Document does not specify a version number of this License, you may choose any version ever published (not as a draft) by the Free Software Foundation.

ADDENDUM: How to use this License for your documents

To use this License in a document you have written, include a copy of the License in the document and put the following copyright and license notices just after the title page:

Copyright © YEAR YOUR NAME. Permission is granted to copy, distribute and/or modify this document under the terms of the GNU Free Documentation License, Version 1.2 or any later version published by the Free Software Foundation; with no Invariant Sections, no Front-Cover Texts, and no Back-Cover Texts. A copy of the license is included in the section entitled "GNU Free Documentation License".

If you have Invariant Sections, Front-Cover Texts and Back-Cover Texts, replace the "with...Texts." line with this:

with the Invariant Sections being LIST THEIR TITLES, with the Front-Cover Texts being LIST, and with the Back-Cover Texts being LIST.

If you have Invariant Sections without Cover Texts, or some other combination of the three, merge those two alternatives to suit the situation.

If your document contains nontrivial examples of program code, we recommend releasing these examples in parallel under your choice of free software license, such as the GNU General Public License, to permit their use in free software.